

This Page Is Inserted by IFW Operations  
and is not a part of the Official Record

## **BEST AVAILABLE IMAGES**

Defective images within this document are accurate representations of the original documents submitted by the applicant.

Defects in the images may include (but are not limited to):

- BLACK BORDERS
- TEXT CUT OFF AT TOP, BOTTOM OR SIDES
- FADED TEXT
- ILLEGIBLE TEXT
- SKEWED/SLANTED IMAGES
- COLORED PHOTOS
- BLACK OR VERY BLACK AND WHITE DARK PHOTOS
- GRAY SCALE DOCUMENTS

**IMAGES ARE BEST AVAILABLE COPY.**

**As rescanning documents *will not* correct images,  
please do not report the images to the  
Image Problem Mailbox.**



## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

<p>(51) International Patent Classification <sup>5</sup> : <b>C08L 93/04</b></p>	<p><b>A1</b></p>	<p>(11) International Publication Number: <b>WO 92/00354</b> (43) International Publication Date: <b>9 January 1992 (09.01.92)</b></p>
<p>(21) International Application Number: <b>PCT/US91/04406</b> (22) International Filing Date: <b>21 June 1991 (21.06.91)</b> (30) Priority data: 545,350                      27 June 1990 (27.06.90)      US 686,033                      16 April 1991 (16.04.91)      US (71) Applicant: <b>FERRO CORPORATION [US/US]; 1000 Lakeside Avenue, Cleveland, OH 44114 (US).</b> (72) Inventors: <b>BURDITT, Neil, A. ; 3289 Kenmore Road, Shaker Heights, OH 44122 (US). ABRAMS, Richard, L. ; 4430 Sir Robert Avenue, North Royalton, OH 44133 (US).</b></p>	<p>(74) Agents: <b>BOISSELLE, Armand, P. et al.; Renner, Otto, Boisselle &amp; Sklar, 1621 Euclid Avenue, 19th Floor, Cleveland, OH 44115 (US).</b> (81) Designated States: <b>AT, AT (European patent), AU, BB, BE (European patent), BF (OAPI patent), BG, BJ (OAPI patent), BR, CA, CF (OAPI patent), CG (OAPI patent), CH, CH (European patent), CI (OAPI patent), CM (OAPI patent), CS, DE, DE (European patent), DK, DK (European patent), ES, ES (European patent), FI, FR (European patent), GA (OAPI patent), GB, GB (European patent), GN (OAPI patent), GR (European patent), HU, IT (European patent), JP, KP, KR, LK, LU, LU (European patent), MC, MG, ML (OAPI patent), MN, MR (OAPI patent), MW, NL, NL (European patent), NO, PL, RO, SD, SE, SE (European patent), SN (OAPI patent), SU, TD (OAPI patent), TG (OAPI patent).</b>  <b>Published</b> <i>With international search report.</i></p>	
<p>(54) Title: <b>LIQUID COLORANT/ADDITIVE CONCENTRATE FOR INCORPORATION INTO PLASTICS</b></p>		
<p>(57) Abstract</p> <p>Liquid concentrates for incorporation into polymeric resins are described that are storage stable and resistant to settling or viscosity changes which comprise: (I) a vehicle comprising: (A) at least one organic rosin material; (B) at least one surfactant; and (II) at least one colorant or additive. The liquid concentrates of the present invention have a body or consistency that allow them to readily flow by gravity and are further characterized by a substantially uniform viscosity. The liquid color concentrates of the present invention also are compatible with a wide variety of plastics and facilitate a reduction in processing deficiencies such as screw-slippage, water carry-over and uneven flow to the pump inlet.</p>		

**FOR THE PURPOSES OF INFORMATION ONLY**

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AT	Austria	ES	Spain	MG	Madagascar
AU	Australia	FI	Finland	ML	Mali
BB	Barbados	FR	France	MN	Mongolia
BE	Belgium	GA	Gabon	MR	Mauritania
BF	Burkina Faso	GB	United Kingdom	MW	Malawi
BG	Bulgaria	GN	Guinea	NL	Netherlands
BJ	Benin	GR	Greece	NO	Norway
BR	Brazil	HU	Hungary	PL	Poland
CA	Canada	IT	Italy	RO	Romania
CF	Central African Republic	JP	Japan	SD	Sudan
CG	Congo	KP	Democratic People's Republic of Korea	SE	Sweden
CH	Switzerland	KR	Republic of Korea	SN	Senegal
CI	Côte d'Ivoire	LI	Liechtenstein	SU	Soviet Union
CM	Cameroon	LK	Sri Lanka	TD	Chad
CS	Czechoslovakia	LU	Luxembourg	TG	Togo
DE	Germany	MC	Monaco	US	United States of America
DK	Denmark				

- 1 -

TITLE: LIQUID COLORANT/ADDITIVE CONCENTRATE FOR  
INCORPORATION INTO PLASTICS

5

Cross-Reference to Related Applications

The present application is a continuation-in-part of the copending parent application, Serial No. 07/545,350 filed June 27, 1990, which is hereby fully incorporated by reference herein.

Technical Field

10

The invention pertains to the field of polymer additives and, more particularly, to methods for modifying polymeric resins and to liquid concentrates used in such methods. The invention is especially concerned with liquid colorant or additive concentrates and with the use of such concentrates in coloring or modifying plastics.

15

Background of the Invention

Traditionally, plastics have been colored by tumble blending the plastic with a solid colorant or additive concentrate such as a coloring powder. The blended material is then fed into an extruder or molding machine where the resin and the colorant are plasticized and the colorant is incorporated into the molten plastic.

20

However, this method has a number of drawbacks. In many cases, a solid colorant or additive concentrate does not uniformly mix with plastic, or it does not disperse uniformly during molding. Further, solid colorants often segregate from the plastic during pre-molding handling because of particle size or specific gravity differences.

25

-2-

In recent years, liquid color concentrates have been developed in an effort to overcome the disadvantages of the traditional solid colorants. Liquid color concentrates are dispersions of a colorant in a carrier vehicle. Being liquids, these colorant concentrates may be accurately metered directly into molding machines or extruders by in-line metering systems, which help overcome many of the disadvantages of solid colorants. They have been used to color thermoplastic resins such as polypropylene, acrylonitrile-butadiene-styrene (ABS), polystyrene, poly(ethylene terephthalate) (PET) and the like. However, liquid color concentrates sometimes lack uniformity and tend to settle or separate into layers after standing at ambient storage conditions.

Several prior art techniques have been developed in order to achieve uniform and storage stable liquid colorant concentrates. One prior art technique requires that vehicles for liquid color concentrates be primarily based on expensive surfactants and wetting agents. For example, U.S. patents 3,956,008 and 4,167,503 disclose colorants having surfactant based vehicles that provide an excellent pigment dispersion media and allow relatively high pigment loadings, at viscosities suitable for use in standard in-line metering equipment.

Although surfactant-based vehicles generally perform satisfactorily, they are subject to several drawbacks. First, the surface active agents used as dispersants are expensive, which can significantly add to the cost of coloring or modifying thermoplastic resins. Secondly, since the surfactants used in the vehicles

-3-

are polar in nature and hygroscopic, the liquid color concentrates are often incompatible with certain plastics. This incompatibility can cause poor mixing in the molder or extruder, resulting in non-uniform color, such as specks of colorant or streaking of colorant.

5 Thirdly, surfactant-based vehicles are often responsible for a decrease in output due to lubrication of the forming equipment, especially at high colorant loadings. For example, extrusion of PET is commonly adversely affected by standard liquid colorant concentrates through "screw slippage" in the molding machine or

10 extruder. A further problem encountered with surfactant-based vehicles occurs in the extrusion of mono-axially oriented polypropylene. A phenomenon known as "water carry-over" occurs when incompatible surfactants move to the surface of the plastic and, being hydrophilic, tend to hold water. This results in an

15 unacceptable amount of water being carried over from the quenching water bath to the take-up equipment. Another problem known as exudation occurs when incompatible vehicles exude, or spew up, to the surface of the molded plastic. Finally, since the colorant and an inert filler such as fumed silica are generally used to control

20 viscosity, this results in a concentrate that is thixotropic, i.e. - one which forms a gel that must be broken up before it can flow.

A second prior art technique developed to achieve uniform and storage stable colorant dispersions involves the use of vehicles

25 not primarily based on surfactants. This technique focuses on thixotropy to maintain a stable dispersion of th colorant or

-4-

additive. As noted above, a thixotropic concentrate is one which will not readily flow under gravity, but which thins to a very low viscosity on stirring or other application of shear.

U.S. patents 4,341,565 and 4,444,714 disclose stable liquid  
5 gel colorants that comprise a suspension of organic pigments in a liquid vehicle that is a plasticizer and which is gelled by an organophilic, expanding lattice clay that is ion exchanged with alkyl ammonium ions. The gel is modified by the addition of certain hydroxyalkyl fatty amines or fatty imidazolinium  
10 quaternaries.

U.S. patents 4,571,416 and 4,624,983 disclose liquid color and additive concentrates that are dispersions of a colorant and/or other additive in a non-surfactant based vehicle. The vehicle includes a low molecular weight addition polymer and an essentially  
15 non-volatile solvent for the low molecular weight addition polymer. The vehicle may also contain a dispersion aid, such as oleic acid, especially when the colorant is an inorganic pigment.

However, the non-surfactant based, thixotropic colorant concentrates are also subject to certain limitations. For example,  
20 the thixotropic concentrates, because of their initially high viscosity, may not flow evenly in a vessel to the pump inlet and are therefore often not readily pumpable for metering a given amount of the colorant concentrate into an extruder or molding machine.

25 Thus, liquid color concentrates available to plastic processors in the past have been inefficient, expensive, and not

-5-

completely satisfactory from the standpoint of achieving color  
uniformity. Accordingly, there continues to be a need for liquid  
colorant/additive concentrates that are: (1) cost effective;  
5 (2) compatible with a wide variety of commercial thermoplastic  
resins; (3) suitable for use in conventional in-line metering and  
blending equipment; and (4) possess good shelf stability and  
uniformity.

U.S. patent 4,188,236 is directed to the production of a non-  
dusting pigment composition by contacting an aqueous dispersion of  
10 an organic pigment with

- (a) a solution or dispersion of an alkaline soluble  
rosin acid derivative;
- (b) a sorbitan ester; and
- (c) a water-insoluble organic carrier having a melting  
15 point below 100°C at a temperature above the melting point of  
both the sorbitan ester and the organic carrier.

U.S. patent 4,415,705 is directed to high-solids pigment  
concentrates containing a dispersing agent formed by reacting a  
poly(lower alkylene)imine with a polyester comprising the reaction  
20 product of hydroxystearic acid, or its oligomer, with tall oil  
rosin. The patent describes a black pigment concentrate that  
includes about 4.8 parts by weight of a maleic-modified pentaeryth-  
ritol ester of rosin which is reported to have a Brookfield  
viscosity of 282 poise initially and 368 poise after three days  
25 aging. The concentrate is reported to flow readily by gravity and  
can be handled by pumps commonly used in the art.



-6-

Summary of the Invention

The invention pertains to liquid concentrates that are storage stable and resistant to settling or viscosity changes which comprise:

- 5 (I) a vehicle comprising:
- (A) at least about 15% by weight of at least one organic rosin material;
  - (B) at least one surfactant; and
- (II) at least one colorant or additive.

10 The liquid concentrates of the present invention have a body or consistency that allow them to readily flow by gravity and are further characterized by a substantially uniform viscosity. The liquid concentrates of the present invention also are compatible with a wide variety of plastics and facilitate a reduction in

15 processing deficiencies such as screw-slippage, water carry-over and uneven flow to the pump inlet.

Brief Description of the Drawing

Fig. 1 is a graphic representation of variations in the log viscosity vs. shear rate for various TiO<sub>2</sub> liquid concentrates.

20

Detailed Description

In one of its aspects, the present invention is directed to liquid color concentrates comprising:

- (I) a vehicle comprising:
- 25 (A) at least about 15 percent by weight of the vehicle of at least one organic rosin material;

**SUBSTITUTE SHEET**

-7-

(B) at least about 0.1 percent by weight of the total concentrate of a surfactant;

(II) up to about 85 percent based on the weight of the total concentrate of at least one colorant or additive.

5       The liquid colorant or additive concentrates of the present invention have a substantially uniform viscosity and are characterized by reduced thixotropy. The viscosity of the liquid concentrates of the invention is usually about 5 to about 1500 poise at 25°C under a shear stress of 12.5 thousand dynes/cm<sup>2</sup>. The liquid  
10 color concentrates of the present invention often possess "length", will flow readily under gravity and can be slightly dilatant at high loadings of colorant or additive.

      The terms "reduced thixotropy" and "substantially uniform viscosity", when used in conjunction with the liquid concentrates  
15 of the present invention, denote a material having a viscosity of no more than about 1500 poise at 0 shear stress and wherein said viscosity changes no more than about an order of magnitude when the shear stress is increased to about 12.5 thousand dynes/cm<sup>2</sup>. For example, a material having an initial viscosity of 1500 poise and  
20 a viscosity under shear stress of about 150 poise would be considered to have a substantially uniform viscosity with reduced thixotropy.

      Length is a property of a material whereby it can be stretched out into a long thread without breaking such as, for example,  
25 honey. In contrast, thixotropic materials tend to be "short", i.e. - they have little or no flowing properties and tend to gel or

**SUBSTITUTE SHEET**

-8-

bunch up. The liquid concentrates of the invention are preferably formulated to possess length.

When the liquid concentrates of the present invention are characterized as flowing readily under gravity it is to be understood that a 200 ml beaker filled with the liquid concentrate would deliver at least 90 percent of its contents within five minutes when inverted.

When the liquid concentrates of the invention are characterized as being slightly dilatant at high loadings of colorant or additive (e.g., greater than about 70% by weight of the concentrate) it is to be understood that the slope of the plot of viscosity (poise) vs. shear rate ( $\text{sec}^{-1}$ ) of such materials will show an increase of up to about 600 poise as shear rate increases from about 5 to about 100 ( $\text{sec}^{-1}$ ).

The liquid concentrates of the present invention employ primarily nonaqueous, rosin based vehicles. In one embodiment, the liquid concentrates of the present invention are substantially free of water (i.e. less than about .01% by weight of the total concentrate).

20

#### Vehicle (I)

##### ORGANIC ROSIN MATERIALS (A)

The organic rosin materials (A) of the vehicle system are known in the art and are available from a variety of commercial sources. Such materials are based upon or derived from gum, wood and/or tall oil rosins which are mainly a mixture of fused-ring, monocarboxylic acids typified by levopimaric acid, abietic acid and

25

-9-

isomers thereof with the empirical formula  $C_{19}H_{29}COOH$ . Other acids such as dihydroabiatic acid ( $C_{19}H_{31}COOH$ ) and dehydroabiatic acid ( $C_{19}H_{27}COOH$ ) may also be present in the rosin materials (A). The rosin materials (A) also include non-polymeric derivatives of rosin acids such as esters, addition salts, hydrogenates, reduction products and the like. Suitable rosin materials generally have viscosity of 10 to about 500 poise at 25°C and are either liquid at 25°C or have a softening point (Hercules drop method) of from about 25°C up to about 200°C and, more often, from about 25°C up to about 80°C. The organic rosin materials (A) generally have an acid number from 0 to about 180 and, more often, from about 5 to about 50.

Representative examples of rosin materials useful in component (A) include those based upon: esters derived from rosin, consisting in main part of pimaric-type acid esters and/or abiatic acid esters, including lower alkyl esters such as methyl, ethyl, propyl, butyl, and the like, glycerol esters, pentaerythritol esters, triethylene glycol esters, phthalate esters, and the like, and their hydrogenates and/or reduction products. Specific examples of commercially available materials useful as component (A) include:

SUBSTITUTE SHEET

Table I

	<u>Rosin material (A)</u>	<u>Acid No.</u>	<u>Softening point °C</u>	<u>Chemical description</u>	<u>Vendor</u>
5	Wood rosin	160	78	90% abietic and/or pimaric-type acids and/or isomers thereof	Hexoles
	Abalyn	6	Liquid	Methylester of wood rosin	"
10	Hercolyn D	6	Liquid	Partial hydrogenate of Abalyn	"
	Abitol	41	Liquid	Partial reductate/hydrogenate of Hercolyn D (dehydroabietyl alcohol)	"
15	Cellolyn 21	8	63	Phthalate ester of Abitol	"
	Ester gum	6	90	Glycerol ester of wood rosin	"
20	Pentalyn A	12	111	Pentaerythritol ester of wood rosin	"
	Staybelite	160	75	Partial hydrogenate (50% H <sub>2</sub> ) of wood rosin	"
25	Foral AX	160	75	Partial hydrogenate (65% H <sub>2</sub> ) of wood rosin	"
	Staybelite ester 3	7	Liquid	Triethylene glycol ester of Staybelite	"
30	Staybelite ester 10	8	83	Glycerol ester of Staybelite	"
	Pentalyn H	12	104	Pentaerythritol ester of Staybelite	"
	Foral 85	9	82	Glycerol ester of Foral AX	"
35	Foral 105	12	104	Pentaerythritol ester of Foral AX	"

Table I (cont.)

	<u>Rosin material (A)</u>	<u>Acid No.</u>	<u>Softening point °C</u>	<u>Chemical description</u>	<u>Vendor</u>
5	Poly-Pale Resin	144	102	90% abietic and/or pimaric-type acids and/or isomers thereof	"
10	Poly-Pale 10 ester	7	114	Glycerol ester of Poly-Pale resin	"
	Pentalyn C	14	133	Pentaerythritol ester of Poly-Pale resin	"
	861 resin	165	75	Vacuum distillate of Poly-Pale resin	"
15	Pentalyn 344	10	108	Pentaerythritol ester of 861 resin	"
	Dymerex	140	148	Vacuum distillate of Poly-Pale resin	"
20	Pentalyn K	13	191	Pentaerythritol ester of Dymerex	"
	Hercoflex 400				"
	Hercoflex 500				"
25	Pentrex G				"
	Synthe copal rosin ester				Raldold

-12-

The vehicle of the liquid concentrate comprises at least about 15 percent by weight based on the weight of vehicle (I) of organic rosin material (A) and may comprise up to about 99.9 percent by weight based on the weight of the vehicle (I) of rosin material (A). In one embodiment, the organic rosin material (A) is present in the concentrate in an amount from about 10% to about 70% by weight of the total concentrate.

#### SURFACTANTS (B)

The liquid color concentrates of the present invention also contain at least one surfactant (B) capable of producing a dispersion of the colorant or additive (II). The surfactant (B) is employed in an amount sufficient to wet the colorant and/or additives and to reduce thixotropy. Surfactants which are nonionic, cationic, anionic, or amphoteric in nature may be used so long as liquid colorants can be produced with high loadings of colorants or additives and good particle dispersion at surfactant concentrations of no more than about 22% by weight of the total concentrate. In one embodiment, the surfactants are employed within the range of about 0.1% to about 10% by weight of the total concentrate and, more often, from about 1% to about 5% by weight of the total concentrate. At concentrations higher than about 22 wt.%, there is a significant reduction in cost-performance.

The surfactant (B) is generally selected based upon the particular colorants and/or additives used in the concentrate. This is because different colorants and additives have different surfactant demands. In general, a particular colorant's or

**SUBSTITUTE SHEET**

-13-

additive's surfactant demand is met at the point wherein a sharp viscosity change (i.e., a difference of at least 100 fold) occurs during preparation of the concentrate. For the purposes of the present invention, when more than about 20 percent by weight (based on the concentrate) of surfactant is required to meet such a demand, this indicates that the surfactant does not have suitable activity.

Suitable surfactants may be selected for inorganic colorants or additives using the following screening procedure:

- 10 (1) Dissolve about 1.5 percent by weight of the candidate surfactant in a mineral oil such as White Oil 7-NF;
- (2) Add about 50 percent by weight of the desired inorganic colorant or additive under stirring until some signs of thixotropy appear;
- 15 (3) Place the dispersion in a mixer at about an impeller tip speed of 5,000 - 6,000 feet per minute for 2 to about 5 minutes and observe the viscosity;
- (4) If the viscosity observed in (3) is lower than about 20 500 poise, additional pigment, up to about 82 percent by weight of the total composition, may be added. If the resultant dispersion becomes thixotropic, the surfactant, although inadequate for high loadings, is adequate for low pigment loadings. If the dispersion remains flow-
- 25 able, the surfactant will be suitable for high pigment loadings;

**SUBSTITUTE SHEET**



-14-

5 (5) If the viscosity observed in (3) is thixotropic, the surfactant composition is adjusted up to about 5.0 percent by weight. If the thixotropy persists, the surfactant is considered unsuitable. If the thixotropy breaks, the surfactant will be suitable for low pigment loadings.

10 (6) Successful surfactants are tested at lower levels (between about 0.5 to about 10 wt%), using a 1:1 mixture of a rosin material (A) such as Hercoflex 500 and mineral oil and up to about 80% by weight of the colorant or additive. Surfactant level is adjusted until a suitable dispersion having the lowest amount of surfactant is found.

15 Suitable surfactants for organic colorants or additives may be selected by using a similar screening procedure. However, since a mixer is not always efficient in dispersing organic colorants and additives, a three-roll mill may be required. This breaks up fine agglomerates, and exposes much more of the colorant or additive surface to the vehicle. About 5-10 wt.% of the candidate surfac-

20 tant is dissolved in mineral oil. About 40 wt.% of the desired organic colorant or additive is then added. Poor surfactants will produce gels (i.e., become thixotropic) at low levels of organic pigments, even under stirring conditions. Such surfactants are generally unsuitable for the purposes of the invention. The

25 samples which do not gel under stirring conditions are passed over a three-roll mill. At the 40 wt.% level, certain organic colorants

**SUBSTITUTE SHEET**

-15-

or additives will produce some thixotropy from the mill. If this occurs, the surfactant level is adjusted until the thixotropy "breaks" (i.e., a viscosity change or difference of at least 100 fold), and the mix flows in a nearly Newtonian fashion. This indicates that the surfactant demand has been met. If the surfactant demand is met by less than about 40 wt%, the candidate surfactant is considered suitable for use in the present invention. A fresh batch is then made up with the final formula, and milled to verify. Also, because of the extra complexity of the three-roll mill, the screening may be done with a 1:1 rosin material (A) (e.g., Hercoflex 500)/mineral oil mix.

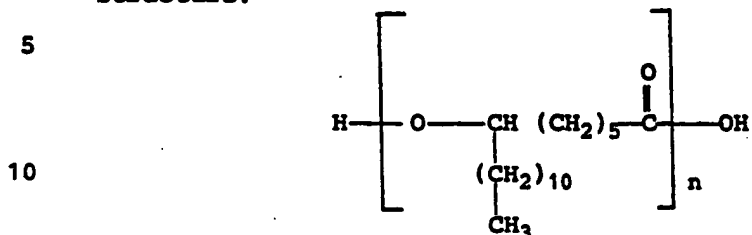
A particularly suitable class of surfactants (B) useful in the practice of the present invention are those based upon polyesters of hydroxy carboxylic acids of the general formula HO-R-COOH where R is a divalent hydrocarbon group having at least about 8 carbon atoms; and polyesters prepared from mixtures of such hydroxy carboxylic acids with a carboxylic acid that is free from hydroxy groups. Representative examples of such hydroxy carboxylic acids include ricinoleic acid, mixtures of 9- and 10-hydroxystearic acid and 12-hydroxystearic acid (HSA). Representative examples of carboxylic acids that are free of hydroxy groups are lauric acid, palmitic acid, stearic acid and oleic acid. Mixtures of 9- and 10-hydroxystearic acid can be obtained by sulphonation of oleic acid followed by hydrolysis. Sources for 12-hydroxystearic acid include commercially available hydrogenated castor oil fatty acid which

SUBSTITUTE SHEET

-16-

contains minor amounts of stearic acid and palmitic acid. Representative examples of suitable polyester surfactants include:

1. Poly-(12-hydroxystearic acid) surfactants of the structure:

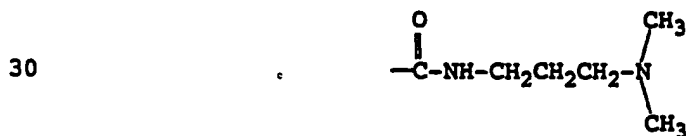


15 wherein (n) represents the degree of polymerization and is usually from about 7 to about 12. In addition to the 12-hydroxystearic acid, such surfactants may contain a lower level of stearic acid. Also, some of the hydroxy endgroups of such poly(12-hydroxystearic acid) surfactants may be capped with regular stearic acid which has

20 no 12-hydroxy group. A specific example of a suitable poly(12-HSA) surfactant is Hypermer® LP-1 from ICI which has a degree of polymerization (n) of about 9.

2. Derivatives of polyester surfactants wherein at least a portion of the acid endgroups are amidized. The polyester

25 surfactants can be amidized with, for example, N,N-dimethyl-1,3-propanediamine to form the following endgroups:



**SUBSTITUTE SHEET**

-17-

A specific example of a suitable amidized polyester is a poly(12-HSA) surfactant wherein about two-thirds of the acid endgroups are amidized with N,N-dimethyl-1,3-propanediamine such as Hypermer® LP-4 from ICI.

5           3. A surfactant obtained from the reaction of a poly(lower alkylene)imine with a polyester. Such adducts are described, for example, in U.S. Patent 4,224,212, which is fully incorporated by reference herein. For example, adducts of poly(12-hydroxystearic acid) and polyethylene imine (PEI) which are formed via both amide and carboxylic acid-amine salt linkages between the carboxylic acid endgroups of the poly(12-HSA) and the amino groups of the PEI have been found to be particularly suitable. The PEI used in preparing such adducts is usually a polyfunctional, randomly branched polymeric amine which contains both primary and secondary amino groups capable of forming amide and salt linkages. A specific example of a suitable poly(12-HSA)-PEI adduct is Hypermer® LP-6 from ICI.

Examples of particular surfactant-additive/colorant combinations useful in the practice of the present invention include:

20

Inorganic Colorants

<u>Colorant</u>	<u>Surfactants</u>
TiO <sub>2</sub>	Hypermer® LP-1 or LP-4
Red Iron Oxide	Hypermer® LP-4
Coated Lead Chromate	Hypermer® LP-4
Ultramarine Blue	Hypermer® LP-4
Mercadium Orange	Hypermer® LP-4
Carbon Black	Hypermer® LP-6

30

**SUBSTITUTE SHEET**

-18-

Organic Colorants and Additives

	<u>Colorant/Additive</u>	<u>Surfactant</u>
5	Azodicarbonamide	Hypermer® LP-6
	Phthalocyanine Blue	Hypermer® LP-6
	Quinacridones	Hypermer® LP-6
	Specialty Azo Reds	Hypermer® LP-6

10

ORGANIC DILUENT (C)

In addition to the above components, it is often necessary that the vehicle portion (I) of the liquid concentrate contain from about 0 to about 80% by weight of a low viscosity organic diluent. As used in conjunction with diluent (C), the term "low viscosity" does not necessarily refer to any particular viscosity range, but is intended to define a material of lower viscosity relative to the more viscous organic rosin based component. The viscosity of (C) is generally below about 10 poise at 25°C. The diluent (C) must be capable of thinning the viscous component (A) at a concentration which will result in a colorant concentrate having the aforementioned properties. Specifically, the organic diluent (C) may comprise from zero to about 80 wt.% of the vehicle (I). Also, suitable diluents are preferably inert to the other components, and have low color, odor, and reactivity at molding or extrusion temperatures.

Examples of such materials include: Mineral oils such as White Oil 7-NF (Technical Products, Inc., Chicago), and the like; mineral spirits such as Odorless Mineral Spirits from Chem Central, and the like; and low molecular weight (below about 1000) polybutenes such

SUBSTITUTE SHEET

-19-

as Amoco Indopol L-14 and the like; and admixtures of one or more of such materials.

Colorants and/or Additives (II)

The term "colorant" when used throughout this specification  
5 and in the appended claims, denotes any conventional inorganic or  
organic pigment, organic dyestuff, or carbon black. Those skilled  
in the art will be aware of suitable inorganic pigments, organic  
pigments and dyestuffs useful in component (II). Such materials  
are described, for example, in Kirk-Othmer Encyclopedia of Chemical  
10 Technology, Third Edition, vol. 6, pp. 597-617 which is incorpo-  
rated by reference herein; examples are:

(1) inorganic types such as titanium dioxide, carbon  
black, iron oxide, zinc chromate, cadmium sulfides, chromium  
oxides, sodium aluminum silicate complexes, such as ultrama-  
15 rine pigments, metal flakes and the like; and

(2) organic types such as azo and diazo pigments,  
phthalocyanines, quinacridone pigments, perylene pigments,  
isoindolinone, anthraquinones, thioindigo, solvent dyes and  
the like.

20 Although principally employed for the preparation of liquid  
colorant concentrates, the vehicle (I) should also be suitable for  
the production of multi-functional dispersions that combine a  
colorant with one or more standard additives. Also, non-pigmented  
additive concentrates which combine one or more additives are  
25 contemplated by the present invention. These additives may include  
UV stabilizers or absorbers, antioxidants, fire retardants, blowing

**SUBSTITUTE SHEET**

-20-

agents, antistats, lubricants, impact modifiers, mold release agents, filler/reinforcing materials (e.g., calcium carbonates), or mixture of two or more of these ingredients. In this way, a single system can impart multiple functional properties to the thermoplastics treated. Hence, unless specifically excluded, when the term colorant is employed for the purpose of describing the concentrate of this invention, it contemplates also additives in general.

Preparation of the colorant concentrates of the invention can be by any of a number of methods known in the art including those set forth in Patton, *Paint Flow and Pigment Dispersion*, John Wiley and Sons (1979) and Weber, ed., *Coloring of Plastics*, SPE Monographs, John Wiley and Sons (1979) which are incorporated by reference herein. In one aspect, the concentrates of the invention are prepared by dissolving the surfactant component (B) in the viscous material (A) at a temperature below about 150°, and usually ambient temperature, in a high speed disperser, such as a high speed rotary mixer (eg. Cowles Dissolver), three roll mill, media mill (sand mill, shot mill, ball mill, etc.) such as those manufactured by Premier Mill. If a diluent (C) is to be employed, it is usually added to the vehicle system at this point. After the surfactant is completely dissolved and the vehicle is homogeneous, the colorant is gradually added under agitation and the mixture is dispersed at high speed until a uniform, stable dispersion is achieved.

In another aspect, the present invention relates to methods of coloring or modifying polymeric resins. The liquid colorant

**SUBSTITUTE SHEET**

-21-

concentrates of this invention are added to the polymeric resin to be colored or modified in amounts up to about 10%, preferably up to about 5%, based on the weight of the resin. The amount used is often referred to as the "letdown" or "letdown ratio" and is  
5 normally dependent on the color intensity desired or required level of additive. Also, admixtures of one or more concentrates may be employed to vary the color or other properties of the resin.

A polymeric resin may be modified or colored by the liquid color concentrates of the present invention by intimate mixing.  
10 "Intimate" mixing means that the mixture is prepared with sufficient mechanical shear and thermal energy to produce a dispersed phase which is finely divided and homogeneously dispersed in the continuous or principal phase. For example, a particularly useful procedure is to intimately mix the polymeric resin with one or more  
15 colorant concentrates using conventional mixing equipment such as a mill, a Banbury mixer, a Brabender torque rheometer, a single or twin screw extruder, continuous mixers, kneaders, etc. The polymers may be intimately mixed in the form of granules, pellets, and/or powder in a high shear mixer. One preferred process for  
20 coloring and/or modifying polymeric resins according to the invention utilizes an in-line metering system for direct incorporation of the concentrate into a polymer resin at the extruder or molding machine. The colorant concentrates may be added in measured amounts to the polymeric resin either by tumbling with the  
25 resin or by pumping the concentrate into the feed throat of a

**SUBSTITUTE SHEET**



-22-

plastic machine or preferably metering into the polymer's melt stream.

The concentrates of the present invention may be used to color or incorporate additives into a wide variety of polymeric resins and find equal application in modification of both thermoplastic and thermoset resins. For example, the color concentrates of the invention are generally compatible with, and may be used with the following resins: polyolefins such as LLDPE, HDPE, LDPE, homopolymer and copolymer polypropylenes, EVA, etc.; styrenics such as crystal polystyrene, HIPS, MIPS, etc.; copolymer resins, resin blends and alloys such as ABS, polyphenylene ether (including PPO blends such as NORYL®), various engineering polymer alloys; polyamide resins, such as Nylon 6, 6/6, 11, 12, copolymers, terpolymers, etc.; acrylic resins, polycarbonates, and other amorphous resins; polyesters, such as PET, PBT, and others - both thermoplastic and curing, as well as resins reinforced with glass, quartz, carbon, aramid, or the like, fibers.

#### EXAMPLES

In order that those skilled in the art will be better able to practice the invention, the following examples are given by way of illustration and not by way of limitation. Unless otherwise noted, all percentages are by weight of the total composition. All temperatures, if not specified, are ambient temperature. Mineral oil, when referred to herein denotes White Oil 7-NF (Technical Products, Inc., Chicago).

**SUBSTITUTE SHEET**

-23-

1. Surfactant ScreeningExample 1

The following formula is made to evaluate Byk Anti-Terra 207, a candidate surfactant:

5	Raven 1170 Carbon Black	30%
	Mineral oil	60%
	Byk Chemie Anti-Terra 207	10%

The mineral oil and surfactant were mixed at low rpms on the high-speed disperser, then the pigment was added gradually under agitation.

After mixing with a high-speed disperser at moderate rpms (under 3000), the mix was a stiff paste. Ten more grams of the surfactant were added, and it was agitated again. The mix remained a stiff paste. Since other candidate materials had produced a fluid mixture under the same conditions, the surfactant was not considered suitable for carbon black in this system.

Example 2

The surfactant of Example 1 is evaluated for Titanium Dioxide dispersions, using the following formula:

20	Kerr-McGee CR-834 TiO <sub>2</sub>	70%
	Mineral oil	28%
	Byk Chemie Anti-Terra 207	2%

The mineral oil and surfactant were mixed at low rpms on the high-speed disperser, then the pigment was added gradually under agitation.

This mixture produced a stiff paste under agitation by the high-speed disperser. A further addition of 2% of the surfactant did not produce a free-flowing liquid. Since other candidate

SUBSTITUTE SHEET

-24-

materials had produced a free-flowing liquid at 1% or less concentration, the material was not considered a successful candidate.

Example 3

5 The following mixture was prepared:

Raven 1170 Carbon Black	30%
Mineral oil	60%
Ethoxylated Sorbitan Trioleate	10%

10 The mineral oil and surfactant were mixed at low rpms on the high-speed disperser, then the pigment was added gradually under agitation.

The mixture produced a thick paste under agitation at low, then high rpms on a high-speed disperser. Ten percent more  
15 surfactant was added, and the mix became free-flowing. Although the mixture became a soft paste on milling, the candidate material was considered to have passed the initial screening.

2. Sample Preparation

20 All samples of colorant concentrate are prepared using a rotary mixer, capable of 20,000 rpm - a one-horsepower Laboratory Dispersator from Premier Mill. The surfactant and all or, in some cases, a portion of the liquid ingredients are first premixed. The colorant is gradually added under constant stirring until all colorant is in the batch. The mixer is then turned up to produce  
25 a smooth vortex, then run at that speed for five minutes.

**SUBSTITUTE SHEET**

-25-

Example 4

A liquid colorant concentrate is prepared from:

	CR-834 Titanium Dioxide	70%
	Hypermer LP-1	1.5%
5	Hercoflex 500	28.5%

The concentrate has a viscosity of about 300 to 500 Poise, and is somewhat dilatant.

Example 5

A liquid colorant concentrate is prepared from:

10	CR-834 Titanium Dioxide	80%
	Hypermer LP-1	1.33%
	Hercoflex 500	13.33%
	Mineral Oil	5.34%

15 The concentrate has a viscosity of about 750 to 865 Poise, and is somewhat dilatant.

Example 6

A liquid additive concentrate is prepared from:

20	Kempore 200 (Olin) Azodicarbonamide (blowing agent)	46.9%
	Indopol L-14 Polybutene	9.1%
25	Mineral Spirits	9.1%
	Sorbitan Trioleate (ethoxylated)	1.0%
30	Abalyn Rosin	33.9%

-26-

Example 7

A liquid color concentrate is prepared from:

	Mapico 297 Iron Oxide	75.0%
	Hercoflex 500	17.2%
5	Mineral Oil	7.3%
	Hypermer LP-4	0.5%

Example 8

A liquid color concentrate is prepared from:

	Monastral Red Y Quinacridone	46.2%
10	Hercoflex 500	28.1%
	Mineral Oil	16.5%
	Hypermer LP-6	9.2%

Example 9

A liquid color concentrate is prepared from:

15	Ultramarine Blue 5151	50.0%
	Hercoflex 500	43.4%
	Mineral Oil	4.6%
	Hypermer LP-4	2.0%

Example 10

20 A liquid color concentrate is prepared from:

	Monarch 120 Carbon Black	42.9%
	Hercoflex 500	36.1%
	Mineral Oil	15.4%
	Hypermer LP-6	5.6%

SUBSTITUTE SHEET

-27-

Example 11

A liquid color concentrate is prepared from:

5	RFC-6 Titanium Dioxide (Tioxide Canada)	70.0%
	Calcium Carbonate	13.5%
	Hercoflex 500	10.5%
10	Mineral Oil	4.5%
	Tinting Pigments	0.6%
15	Hypermer LP-4	0.9%

Example 12

A liquid additive concentrate is prepared from:

	Kempore 200 blowing agent	66.7%
	Hercoflex 500	21.0%
20	Mineral oil	9.0%
	Hypermer LP-6	3.3%

The viscosity of this material ranged from 500 to 1400 poise, and flowed readily under gravity.

Example 13

25 A liquid additive concentrate is prepared from:

	Kempore 200 blowing agent	50.0%
	Hercoflex 500	33.25%
	Mineral oil	14.25%
	Hypermer LP-6	2.50%

30 The viscosity of this material ranged from 26 to 65 poise, and it flowed readily.

-28-

Example 14

A liquid concentrate is prepared from:

5	RFC-6 Titanium Dioxide (Tioxide Canada)	62.36%
	Columbian Chemicals Mapico TAN 20 (magnesium ferrite)	0.79%
10	Hercoflex 500	20.3%
	Hypermer LP-4	1.0%
	Mineral Oil	8.7%
15	Calcium Carbonate	5.3%
	Dispersion of Example 9	1.55%

20

Example 15

	Methyl Ester of Rosin	65.84%
25	Hypermer LP-4	1.50%
	Hypermer LP-6	0.15%
	Glycerol Mono-oleate	5.00%
30	Titanium Dioxide White	0.67%
	Coated Lead Molybdate Orange	2.67%
	Magnesium Ferrite Tan	21.34%
35	C.I. Solvent Yellow 93	2.33%
	Carbon Black	0.50%

40

The above brown coloring material was mixed on a high-speed disperser and passed over a three-roll mill. It was then molded in High Impact Polystyrene on a production injection molding machine with a letdown of 1.6%. The part was free of the streaks which had been common with other coloring methods on the same tooling.

45

Example 16

SUBSTITUTE SHEET

-29-

	Methyl Ester of Rosin	43.54%
	Hypermer LP-4	1.50%
	Carbon Black	0.41%
	Titanium Dioxide White	13.80%
5	Chrome Antimony Titanate	40.75%

The above gold coloring material was mixed on a high-speed disperser and passed over a three-roll mill. It was then molded in High Density Polyethylene at a 0.5% letdown ratio on a production injection molding machine. The part was free of the streaks which had been common with other coloring methods on the same tooling with the same colorant combination.

Example 17

	Refined Soya Oil	16.26%
	Methyl Ester of Rosin	38.00%
15	Hypermer LP-4	1.72%
	Hypermer LP-6	1.72%
	Titanium Dioxide White	21.94%
	Carbon Black	13.22%
	Phthalocyanine Blue	4.69%
20	Cadmium Sulfide/Selenide Red	2.45%

The above materials were mixed on a high-speed disperser and passed twice over a three-roll mill. The resulting product was then incorporated into a polyester fiber at letdowns ranging from 1 to 5.9%.

**SUBSTITUTE SHEET**



-30-

Comparative Example 18

A liquid color concentrate is prepared from:

	CR-834 Titanium Dioxide	60%
	H-1900 Polybutene	19%
5	Mineral oil	19%
	Oleic acid	2%

Example A

Certain liquid concentrates prepared in the above examples are evaluated for viscosity/thixotropy by the procedure shown below.

10 The color concentrates prepared in Examples 4, 5 and 18 were measured on a Carri-Med cone and plate viscometer. The shear stress was increased gradually from 0 to about 25000 dynes/cm<sup>2</sup> value, and viscosity was recorded as a function of shear rate (sec<sup>-1</sup>). It can be seen from Figure 1 that Comparative Example 18  
15 has an extremely high initial value (about 70,890 poise). It was, in fact, a gel - which did not flow when the container was held inverted. With increased shear, the Example 18 material dropped dramatically in viscosity, to a level of about 60 poise. Example 3 had only a slight initial drop in viscosity, then a gradual  
20 increase to about 160 poise. Example 4 also had only a slight initial drop in viscosity, then increased to about 850 poise.

Both Example 4 and Example 5, at 70% and 80% Titanium Dioxide respectively, flowed readily under gravity; and their initial viscosities were well below the Comparative Example 18 at 60%  
25 Titanium Dioxide.

SUBSTITUTE SHEET

-31-

Examples 7 and 8 - 75% Iron Oxide Red and 45.2% Quinacridone Red respectively - both flowed freely under gravity, then thickened under increasing shear.

5 Examples 9 and 10 - 50% Ultramarine Blue and 42.9% Carbon Black respectively - both flowed freely under gravity. The Ultramarine Blue was essentially Newtonian in behavior, while the Carbon Black displayed some dilatency.

10 Example 11 - 83.5% total solids - 70% TiO<sub>2</sub>, 13.5% CaCO<sub>3</sub> in a rosin vehicle showed relatively low viscosity and moderate dilatency.

Example B

A vehicle of 69.75% Hercoflex 500 and 29.75% Mineral oil, with 0.5% Solvent Red 111 incorporated as a visual tracer, was added to the following resins: High Density Polyethylene (HDPE),  
15 Homopolymer Polypropylene (HOMO.PP), Copolymer Polypropylene (COPOLY PP), Crystal Polystyrene (GPPS), High Impact Polystyrene (HIPS), and Polycarbonate - at a level of 4%. This is a level well above typical usage rates for pure vehicle. The polymer and vehicle were mixed by hand in a polyethylene bag, and injection  
20 molded into 1/8" tensile bars for tensile testing, and into square plaques for visual inspection and Gardner Impact tests.

There was no screw slippage observed in the above moldings; the samples showed no visual sign of nonuniformity or delamination. The physical test results were as follows:

-32-

Table 1

	<u>Resin</u>	<u>Wt% Vehicle</u>	<u>Yield Strength<sup>1</sup></u>	<u>Gardner Impact<sup>2</sup></u>
	HIPS	--	4147(44)	10(7)
	HIPS	4	3353(90)	54(43)
5	Homo. PP	--	4513(103)	60(26)
	Homo. PP	4	4539(70)	149(9)
	COPOLY PP	--	3385(63)	175(78)
	COPOLY PP	4	3423(10)	176(17)
	HDPE	--	3582(41)	203(35)
10	HDPE	4	3047(124)	205(59)
	Polycarbonate	--	8955(157)	>320
	Polycarbonate	4	9602(155)	>320
	GPSS	--	--	<2
	GPSS	4	--	<2

15

<sup>1</sup> PSI ASTM D 638, parenthetical number is standard deviation  
<sup>2</sup> in-lbs (J) ASTM D 3029 parenthetical number is standard deviation

20

25 While the invention has been explained in relation to its preferred embodiments, it is to be understood that various modifications thereof will become apparent to those skilled in the art upon reading the specification. Therefore, it is to be understood that the invention disclosed herein is intended to cover such modifications as fall within the scope of the appended claims.

SUBSTITUTE SHEET

-33-

What is claimed is:

1. A liquid concentrate for incorporation into a polymeric resin which comprises:

(I) a vehicle comprising:

5 (A) at least about 15% by weight of at least one organic rosin material; and

(B) at least one surfactant; and

(II) at least one colorant or additive.

10 2. A liquid concentrate for incorporation into a polymeric resin which comprises:

(I) a vehicle comprising:

(A) at least about 15% by weight of at least one organic rosin material;

15 (B) at least about 0.1% by weight of at least one surfactant; and

(C) from 0% up to about 80% by weight of at least one organic diluent having a viscosity under shear up to about 10 poise; and

20 (II) Up to about 85% based on the weight of the total concentrate of at least one colorant or additive;

wherein said concentrate flows readily by gravity and is characterized by a viscosity at 25°C under a shear stress of 12.5 thousand dynes/cm<sup>2</sup> of about 5 to about 1500 poise.

25 3. The concentrate of claim 2 wherein component (A) comprises a lower alkyl ester of an abietic acid based rosin.

**SUBSTITUTE SHEET**

-34-

4. The concentrate of claim 3 wherein said lower alkyl ester comprises the methyl ester of abietic acid.

5. The concentrate of claim 2 wherein component (A) comprises a glycerol, pentaerythritol, triethylene glycol or phthalate ester of an abietic acid based rosin.

6. The concentrate of claim 2 wherein component (B) comprises a poly(12-hydroxystearic acid) based surfactant.

7. The concentrate of claim 6 wherein at least a portion of the acid end groups of said poly(12-hydroxystearic acid) are amidized.

8. The concentrate of claim 7 wherein said acid end groups are amidized with N,N-dimethyl-1,3-propanediamine.

9. The concentrate of claim 2 wherein said surfactant is an adduct of poly(12-hydroxystearic acid) and a polyethyleneimine.

10. The concentrate of claim 2 wherein component (II) comprises one or more colorants, U.V. stabilizers, antioxidants, fire retardants, blowing agents, antistats, lubricants, impact modifiers, mold release agents, or fillers.

11. A method for modifying a polymeric resin which comprises intimately blending therewith an effective amount of a liquid concentrate comprising:

(I) a vehicle comprising:

(A) at least about 15% by weight of at least one organic rosin material;

(B) at least about 0.1% by weight of a surfactant;  
and

**SUBSTITUTE SHEET**

-35-

(C) from 0 up to about 85% by weight of a low viscosity organic diluent; and

(II) up to about 85 percent based on the weight of the total concentrate of at least one colorant or additive.

5 12. The method of claim 7 wherein component (A) comprises a lower alkyl ester of an abietic acid based rosin.

13. The method of claim 12 wherein said lower alkyl ester comprises the methyl ester of abietic acid.

10 14. The method of claim 11 wherein component (A) comprises a glycerol, pentaerythritol, triethylene glycol or phthalate ester of an abietic acid based rosin.

15 15. The method of claim 7 wherein component (B) comprises a poly(12-hydroxystearic acid) based surfactant.

16 16. The method of claim 11 wherein at least a portion of the acid end groups of said poly(12-hydroxystearic acid) are amidized.

17. The method of claim 16 wherein said acid end groups are amidized with N,N-dimethyl-1,3-propanediamine.

18. The method of claim 11 wherein said surfactant is an adduct of poly(12-hydroxystearic acid) and a polyethyleneimine.

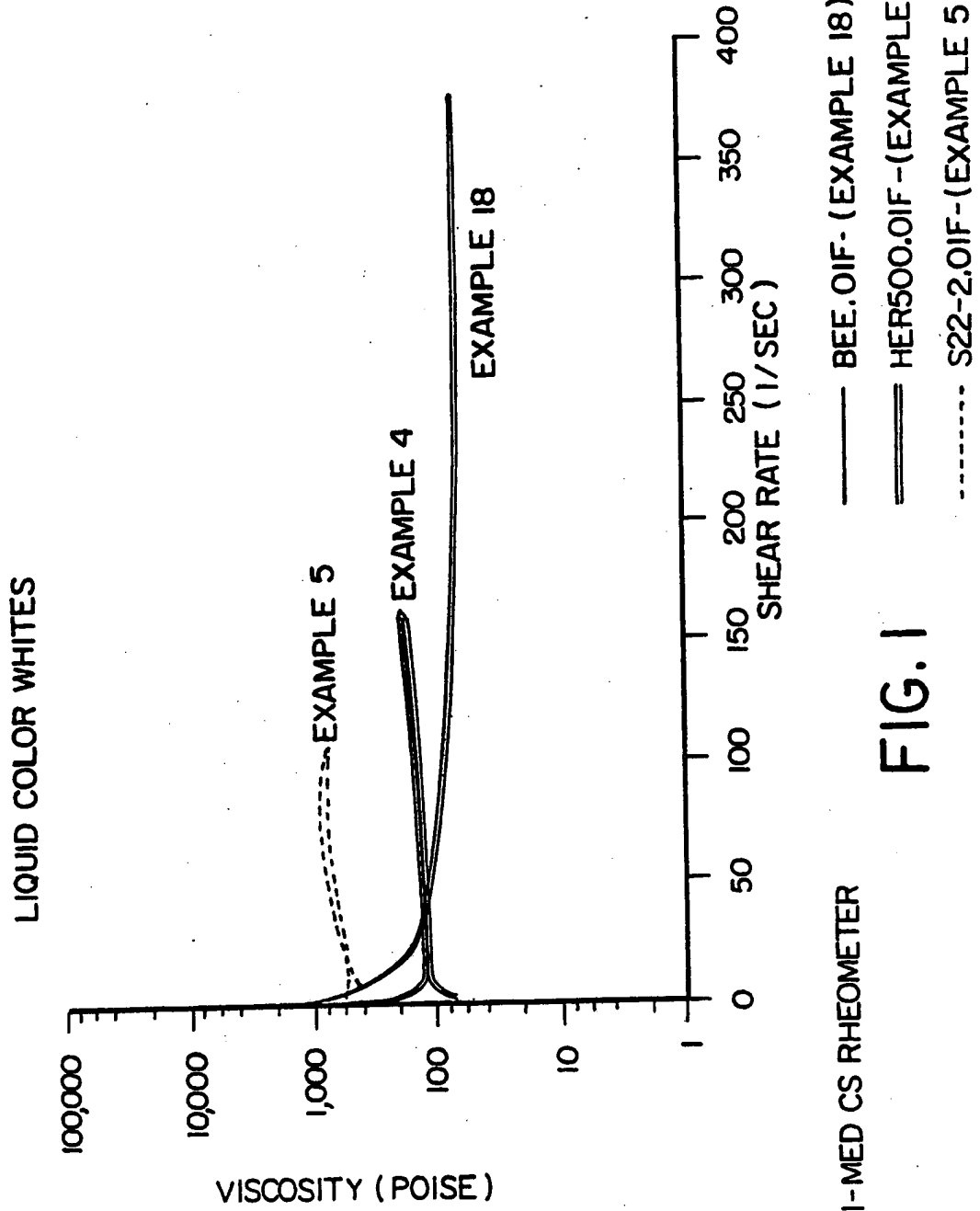
20 19. The method of claim 7 wherein said diluent comprises a mineral oil.

25 20. The method of claim 11 wherein said additive is selected from the group comprising colorants, U.V. stabilizers, antioxidants, fire retardants, and blowing agents, antistats, lubricants, impact modifiers, and mold release agents.

-36-

21. The method of claim 11 wherein said concentrate has a viscosity under shear of 12.5 thousand dynes/cm<sup>2</sup> is between about 5 and about 1,500 poise at 25°C.

**SUBSTITUTE SHEET**




CARRI-MED CS RHEOMETER

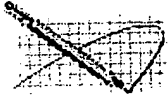
FIG. 1



# INTERNATIONAL SEARCH REPORT

International Application No **PCT/US91/04406**

<b>I. CLASSIFICATION F SUBJECT MATTER</b> (if several classification symbols apply, indicate all) <sup>2</sup>		
According to International Patent Classification (IPC) or to both National Classification and IPC		
IPC (5): <b>C08L 93/04</b>		
U.S. Cl: <b>524/270</b>		
<b>II. FIELDS SEARCHED</b>		
Minimum Documentation Searched <sup>4</sup>		
Classification System <sup>1</sup>	Classification Symbols	
U.S.	524/270	
Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in the Fields Searched <sup>5</sup>		
<b>III. DOCUMENTS CONSIDERED TO BE RELEVANT</b> <sup>13</sup>		
Category <sup>6</sup>	Citation of Document, <sup>14</sup> with indication, where appropriate, of the relevant passages <sup>17</sup>	Relevant to Claim No. <sup>14</sup>
Y	US, A, 4,634,471 (FOYE ET AL) 06 JANUARY 1987 See claims 1-13.	1-21
Y	US, A, 4,885,033 (BLACKBURN ET AL) 05 DECEMBER 1989 See column 4, lines 54-58.	1-21
<p><sup>13</sup> Special categories of cited documents:</p> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p> <p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</p> <p>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.</p> <p>"&amp;" document member of the same patent family</p>		
<b>IV. CERTIFICATION</b>		
Date of the Actual Completion of the International Search <sup>8</sup>	Date of Mailing of this International Search Report <sup>9</sup>	
26 AUGUST 1991	<b>24 SEP 1991</b>	
International Searching Authority <sup>1</sup>	Signature of Authorized Officer <sup>10</sup>	
ISA/US	 W. C. Cheng	



**Kevin M Faulkner**

**07/17/02 04:17 PM**

**To: Lisa L Schoppa/B/Baytown/ExxonMobil@XOM**  
**cc:**  
**Subject: patent copies**

**also get**

**WO 92/00354**

**thanks!**

**Kevin M. Faulkner**  
**Attorney**  
**ExxonMobil Chemical Company**  
**Baytown Polymers Center/Law Technology**  
**phone: 281-834-2677**  
**fax: 281-834-2495**